# **BUTYRIC ACID**



## 1. PERFORMANCE

1) Measuring range 3-60 ppmNumber of pump strokes  $1 (100 \text{m} \ell)$ 

2) Sampling time : 1.5 minutes/1 pump stroke

3) Detectable limit : -4) Shelf life : 3 years
5) Operating temperature :  $15 \sim 25 \,^{\circ}\text{C}$ 

6) Reading : Graduations printed on the tube are calibrated by Acetic acid at 1 pump stroke

and Butyric acid concentration is determined by using a conversion chart at

1 pump stroke.

7) Colour change : Pale pink → Yellow

## 2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

## 3. CHEMICAL REACTION

By reacting with alkali, PH indicator is discoloured. CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>COOH + NaOH → C<sub>3</sub>H<sub>7</sub>COONa + H<sub>2</sub>O

## 4. CALIBRATION OF THE TUBE

VAPOUR PRESSURE METHOD

## 5. INTERFERENCE AND CROSS SENSITIVITY

Substance	ppm	Interference	Coexistence
Sulphur dioxide		Similar stain is produced.	Higher readings are given.
Nitrogen dioxide	300	"	The top of discoloured layer becomes unclear.
Hydrogen chloride		Pink stain is produced.	Higher readings are given.
Chlorine		Blueish yellow stain is produced.	"
Acetic acid		Similar stain is produced.	"

